# 國立臺北科技大學 102 學年度碩士班招生考試

系所組別:3722 有機高分子研究所乙組

第二節 材料科學與工程 試題 (選考)

第一頁,共一百

### 注意事項:

- 一. Explain following term (use graph or give an example): 【30%, 每小題 6 分】

- 1. Interstitial solid solution
- 2. Frenkel imperfection
- 3. Extrinsic semiconductor (n-type and p-type)
- 4. Yield strength (0.2 percent offset yield strength)
- 5. Craze (in brittle or ductile fracture of polymeric materials?)

#### 二. 【14%】

Calculate the intrinsic electrical conductivity of InSb at 27°C and at 60°C. [Eg = 0.17 eV; at 27°C:  $\mu_n = 8.0 \text{ m}^2/(\text{V} \cdot \text{s})$ ,  $\mu_p = 0.045 \text{ m}^2/(\text{V} \cdot \text{s})$ ;  $n_i = 1.35 \cdot 10^{22} \text{ m}^{-3}$ 

## [10%]

Draw a typical creep curve for a metal under constant load and at a relatively high temperature, and indicate on it all three stages of creep. Give explanation.

## 四. 【12%, 每小題 6 分】

X rays of an unknown wavelength are diffracted by a gold sample. The 2 theta angle was 64.582° for the {220} planes.

- a) Determine the inter-planar d-spacing of {220} plane of gold.
- b) Determine the wavelength of the X rays used?

(The lattice constant of gold = 0.40788 nm; assume first-order diffraction, n = 1.).

## 五. 【12%】

Which equipment can you use to observe dislocations (line defects) and grain boundaries, respectively? Explain why or how to use the equipment for the observation of these defects.

## 六. 【10%】

The diffusivity of manganese atoms in the FCC iron lattice is 1.50 · 10<sup>-14</sup> m<sup>2</sup>/s at 1300°C and 1.50 · 10<sup>-15</sup> m<sup>2</sup>/s at 400°C. Calculate the activation energy in kJ/mol for this case in this temperature range.  $R = 8.314 \text{ J/(mol \cdot K)}$ .

#### 七. 【12%, 每小題4分】

A hydrogen atom exists with its electron in the n = 4 state. The electron undergoes a transition to the n = 3 state.

Calculate (a) the energy of the photon emitted, (b) its frequency, and (c) its wavelength in nanometers (nm).  $1 \text{ eV} = 1.6 \cdot 10^{-19} \text{ J}$  $h = 6.63 \cdot 10^{-34} \, J \cdot s$